



Selective laser melted TiB₂/Ti6Al4V graded materials and first-principle calculations

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ABSTRACT

Selective laser melting (SLM), as a rapidly developed layer-by-layer 3D printing technology, has been demonstrated that it can be used for additive manufacturing of multi-materials. In this work, TiB₂/Ti6Al4V multi-materials were manufactured by SLM. The results showed that interfacial microstructures consisting of un-melted TiB₂ powders, columnar TiB crystals, and acicular TiB crystals were formed in the molten pool. Meanwhile, a variation nanohardness was developed with chemical compositions at the interfacial regions, owing to an in-situ reaction of Ti and TiB₂. The first-principle calculations were used to reveal the axial growth of TiB crystals as well as the different hardness of this system.

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1. Introduction

Selective laser melting (SLM), as a rapidly developed layer by layer additive manufacturing technique, is capable to produce high densification and complex parts by using of metal, ceramic, or composite materials [1,2]. For many practical engineering applications, multi-materials SLM is indispensable due to the requirement of different material properties at different regions. Tan et al. have clearly revealed the feasibility of combining different metals with good metallurgical bonding using SLM [3]. Compared with metal-metal multiple materials, there are more challenges for SLM of metal-ceramic multiple materials. These are involved with the extremely different bond strength and thermal expansion coefficient between metal and ceramic, leading to thermal stress and even delamination during non-equilibrium solidification of SLM [4].

To overcome the above constraints, fabrication of functional graded materials (FGM) through good interfacial bonding between metal and ceramic is feasible. This method can avoid residual stress concentration and fracture propagation by reducing the saltation of different properties between neighboring layers [5,6]. Shishkovsky et al. [7] have successfully fabricated functional

graded specimens and realized a hardness gradient via SLM using the compositionally grade powders (Ti+(10, 15 or 20 vol.%) TiB₂). The nanohardness of SLMed Ti6Al4V with 3 wt% TiB₂ can reach to 6.00 GPa due to the reinforcement effect of needle-like TiB [8]. Nevertheless, it is difficult to make a larger range of hardness difference for multi-materials SLM of pure TiB₂ and Ti6Al4V powders instead of composite powders. Interfacial reaction and microstructure are vital to the connection characteristic of adjacent layers. In this work, graded materials were prepared by SLM of TiB₂/Ti6Al4V multi-materials. The microstructure and nanohardness at the interface were investigated. Additionally, first-principle calculations were used to explain the axial growth of TiB and different hardness of this system.

2. Materials and methods

The SLM equipment was composed of an IPG YLR-500 Ytterbium fiber laser with a 1070 ± 10 nm wavelength. Fig. 1a shows the SLM process of Ti6Al4V and TiB₂ powders. Cubic samples were fabricated layer by layer on a Ti6Al4V substrate. Based on our previous research [9], the laser power (*P*), scanning speed (*v*) and hatch spacing of Ti6Al4V layer were 250 W, 600 mm/s, and 50 μm, respectively. While for the TiB₂ layer, the respective laser scanning speed and hatch spacing were set as 80 mm/s and 120 μm. The laser power varied from 200 W to 500 W with an interval of 50 W. The starting powders of Ti6Al4V (99.7% purity,

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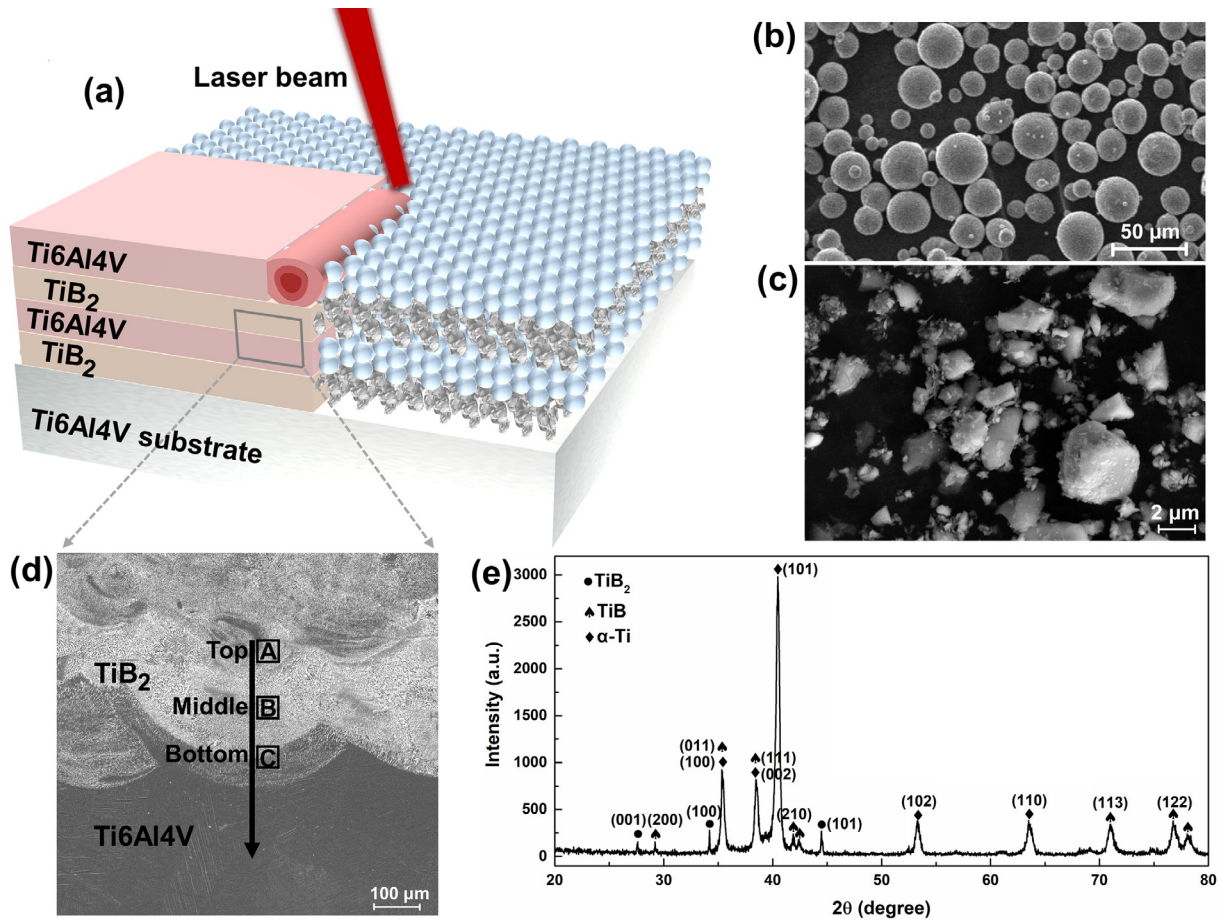


Fig. 1. (a) Schematic of SLM process with $\text{TiB}_2/\text{Ti6Al4V}$ multi-materials; FE-SEM images of (b) TiB_2 and (c) Ti6Al4V powders; (d) cross-section of SLMed $\text{TiB}_2/\text{Ti6Al4V}$ multi-materials under $v = 80$ mm/s, $P = 400$ W; (e) XRD pattern of SLMed $\text{TiB}_2/\text{Ti6Al4V}$ multi-materials taken from the interface.

D_{50} : 24.06 μm) with a spherical shape and TiB_2 (98.9% purity, D_{50} : 2.21 μm) with an irregular shape are shown in Fig. 1b and c.

Phase identification was studied by X-ray diffraction (XRD, Cu $K\alpha$). Microstructures were characterized by field emission scanning electron microscope (FE-SEM, Hitachi, Japan). The chemical compositions were determined by an energy dispersive spectrometer (EDS). Besides, nanohardness test was measured on the polished samples with a force of 20 mN and holding time of 10 s by a nanoindentation tester (Agilent Nano Indenter G200, USA). The first-principle calculations were performed using the VASP code. Considering the occurrence of α -Ti in SLM-processed Ti6Al4V samples [8], crystal structure of pure Ti crystal was used to simplify the complex crystal of Ti6Al4V .

3. Results and discussion

For a low laser power of 200 W and 250 W, many pores appeared at the cross-section of SLM-processed $\text{TiB}_2/\text{Ti6Al4V}$ multi-materials under v of 80 mm/s. As the laser power increased to 350 W, samples with good interfacial bonding were obtained because of enough energy input. Hence, the SLM-processed sample at a laser power of 400 W was taken as an example to investigate interfacial microstructure and property (Fig. 1d). The XRD pattern of sample taken from the interface showed the presence of α -Ti, TiB and TiB_2 diffraction peaks (Fig. 1e). Recent researches on TiB_2 reinforced Ti6Al4V composites illustrated that reaction between Ti and TiB_2 can proceed as follows: $\text{Ti} + \text{TiB}_2 = 2\text{TiB}$ [8,10], that is to say, TiB can be formed by a reaction of Ti and TiB_2 under high

laser radiation. This explains the appearance of TiB phase in the SLM-processed samples.

Fig. 2a shows the SEM micrographs of three different regions in molten pool: Top (A), middle (B), and bottom (C) as shown in Fig. 1d. From Fig. 2b, it can be found that the atomic ratio of Ti and B at point 1 is 1:2. It means that in the region A, a small amount of TiB_2 particles remained in an irregular shape are consistent with the weak TiB_2 peaks of XRD pattern. Columnar and fine acicular crystals were observed in the middle and bottom of molten pool, respectively. An atom ratio of Ti and B with 1:1 for points 2 and 3 has proved to be TiB crystals. The different diameters of TiB crystals are caused by a large temperature gradient and different cooling rate during high-power laser melting. Finer crystal is obtained from the bottom of molten pool with higher cooling rates, attributed to shorter growth time of crystal. In addition, the temperature is the highest at the pool center and decreases gradually with distance away from the center. Thus, the columnar crystals in the region A with lower cooling rate, the acicular crystals in the region B with higher cooling rate and un-melted TiB_2 in the region C with lower temperature can be observed. From the Fig. 2b and c, it can be seen that Al and V elements also existed in the top and middle of the molten pool, which indicated that elements of Ti, Al, and V could diffuse into the interface. Considering the α -Ti generated during laser melting of Ti6Al4V [8], TiB crystals can be produced by a chemical reaction of Ti with TiB_2 at the interface, which led to the formation of gradient interface between the TiB_2 and Ti6Al4V . Based on the results of EDS line scanning, the content of B in the top of molten pool was obviously high at a

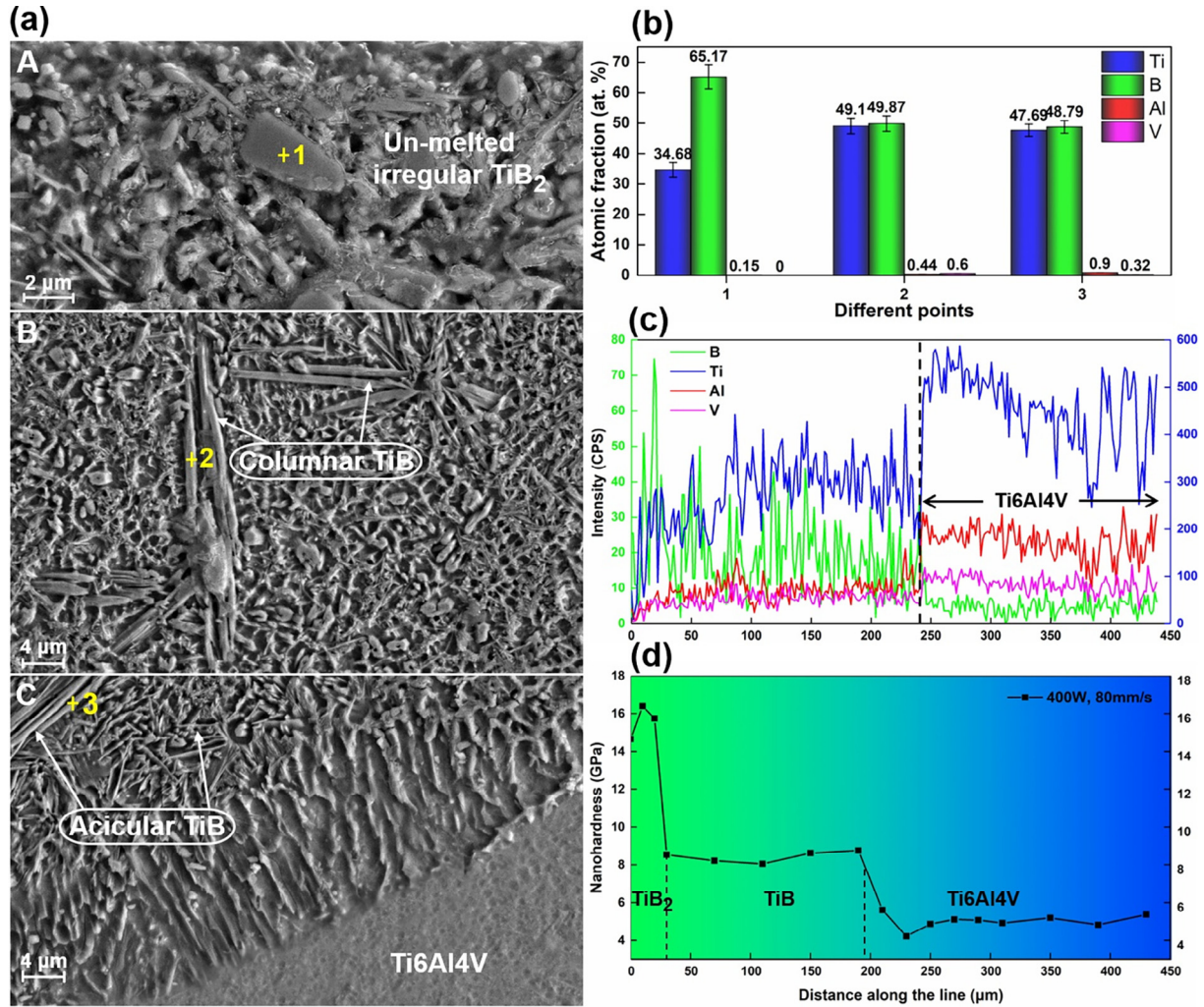


Fig. 2. (a) FE-SEM micrographs showing the different regions of A, B and C of the molten pool in Fig. 1d; (b) chemical compositions by EDS analysis: atomic fraction of Ti, B, Al and V elements at three different positions in Fig. 2a; (c) EDS line-scan analysis and (d) nanoindentation along the line in Fig. 1d.

depth of 0–30 μm and its content gradually stabilized at about 30–240 μm, then started to decline. This suggests that boron diffusion is hard to traverse the pool boundary into Ti6Al4V. However, the increasing intensity of Ti, Al, V at a depth of 250 μm indicate the reaching of Ti6Al4V layer. In general, the TiB crystals in the transition region alleviate the residual stress due to gradient compositions.

Nanohardness measurements were conducted on the interfacial region of TiB₂/Ti6Al4V and the results are shown in Fig. 2d. Nanohardness with an approximate value of 4.8 GPa for Ti6Al4V layer is comparable to reported values [11,12]. According to the gradient change of nanohardness across the interfacial region, it is noted that TiB makes a good transition between the TiB₂ and Ti alloy. Therefore, TiB₂/TiB/Ti6Al4V graded materials promote the formation of good bonding between dissimilar materials. This graded materials additive manufacturing is beneficial to combine the hard ceramic with ductile metal into an organic whole effectively.

The theoretical calculations were performed to investigate the physical origins of different hardness between TiB₂, TiB, and Ti6Al4V. Fig. 3a shows that α-Ti are composed of metallic bond ($d_{Ti-Ti} = 2.86 \text{ \AA}$). From the supercell, both TiB and TiB₂ are mainly consisted of B-B covalent bonds, and the covalent bond length of

TiB ($d_{B-B} = 1.81 \text{ \AA}$) is larger than TiB₂ ($d_{B-B} = 1.75 \text{ \AA}$). Therefore, the covalency of TiB is weaker than that of TiB₂. As the hardness of a material is a reflection of the internal bonding strength, and the impacts of several chemical bonds are covalent bond > ionic bond > metal bond. Thus, the hardness order of these three materials is TiB₂ > TiB > Ti (Ti6Al4V). Apart from the crystal structure calculation, the electronic structure has been also studied. The charge distributions of TiB and TiB₂ are shown in Fig. 3b, zigzag chain of TiB and hexagonal network structure of TiB₂ demonstrate the delocalized π-bonds between the boron atoms. In this situation, TiB structure characterized by zigzag chains of boron atoms is parallel to the [0 1 0] direction, this anisotropic crystal structure may be responsible for an axial growth as observed in Fig. 2a. From Fig. 3c, it can be seen that α-Ti exhibits a good metallicity due to local electronic density of states around the fermi level. The electronic structures of both TiB and TiB₂ are governed by the strong hybridization between the Ti-3d and B-2p states. In particular, the appearance of the pseudo-gap (yellow region) around the fermi level also lead to increased stability of TiB₂. Therefore, the structure stability is in an order of TiB₂ > TiB > Ti6Al4V. Hardness and elasticity of these materials with distinct bonding may show a similar gradient tendency, which contributes to less residual stress and good interfacial bonding.

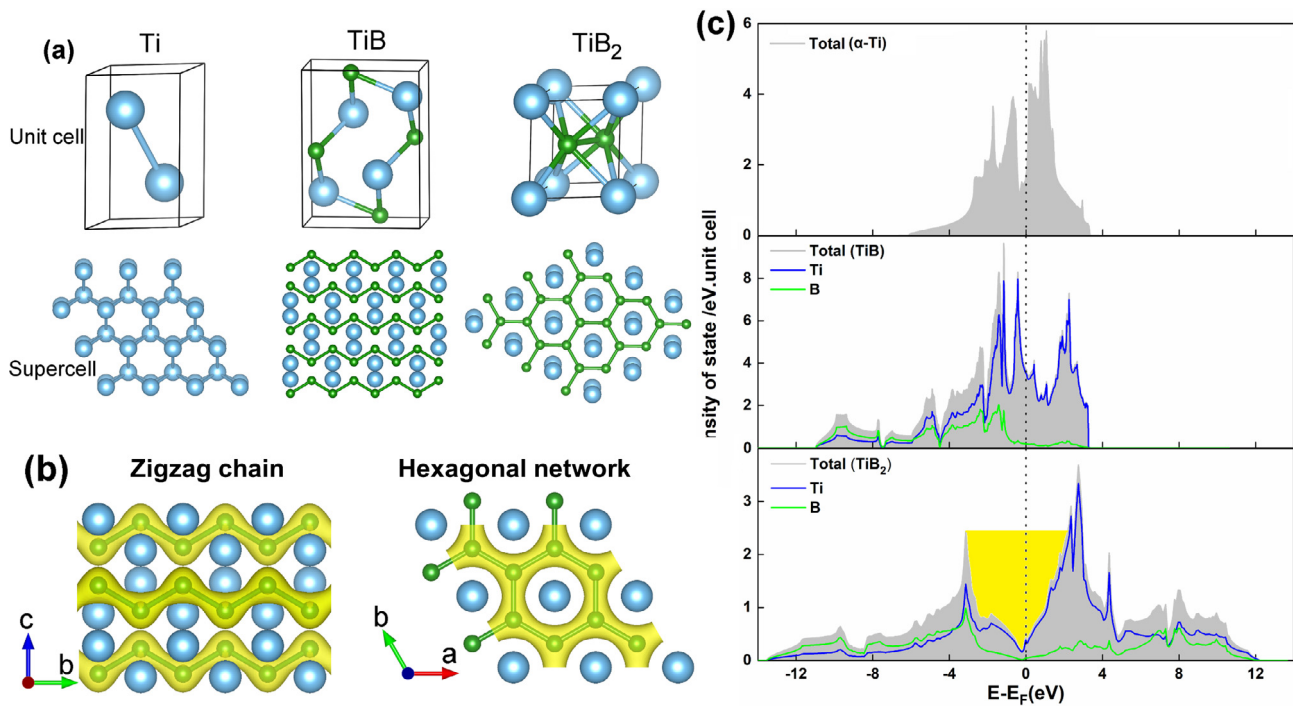


Fig. 3. (a) Unit cell and supercell of Ti, TiB and TiB₂; (b) charge distribution of TiB and TiB₂; (c) partial density of states of the system.

4. Conclusions

TiB₂/TiB/Ti6Al4V graded materials have been achieved by SLM of TiB₂/Ti6Al4V multi-materials. TiB was formed by an in-situ reaction of Ti with TiB₂ and the gradual variations of chemical compositions caused a variation in nanohardness along the interface. The structural stabilities of TiB₂, TiB, and Ti6Al4V were also studied by first-principle calculations. The hardness gradient of TiB₂ > TiB > Ti6Al4V was demonstrated by our theoretical and experimental investigations. These findings demonstrate that graded materials additive manufacturing is beneficial to fabricate metal-ceramic part successfully.

Declaration of Competing Interest

The authors have declared that no conflict of interest exists.

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